## Norzooanemonin in the hydroid Tubularia larynx1

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Summary. The hydrochloride of 1,3-dimethyl imidazole-4-carboxylic acid (norzooanemonin), has been isolated from the hydroid Tubularia larynx and its structure determined by X-ray analysis.

In the process of purifying the sperm attractant(s) of the hydroid,  $Tubularia\ larynx$ , we have isolated trigonelline and homarine (N-methyl nicotinic acid betaine and N-methyl  $\alpha$ -picolinic acid betaine, respectively)  $^2$ . While we have not yet identified the sperm attractant, we now report the isolation and X-ray analysis of norzooanemonin hydrochloride (I). The betaine from I may be regarded as the nor-derivative of zooanemonin, the homologous dimethyl imidazole acetic acid betaine, which was isolated initially by Ackerman  $^3$  from a sea anemone and has since been encountered in other marine invertebrates  $^{4,5}$ .

The crystalline hydrochloride (m.p. 216–219°C, from methanol: ethanol, 4:1; Weinheimer<sup>6</sup> 213–217°C) was isolated in low yield (8 mg) by ion exchange and preparative paper chromatography of a water soluble fraction from the original ethanol extract.

Due to the paucity of material an X-ray analysis was done. The crystals of the hydrochloride I were orthorhombic with a = 6.403(1), b = 10.721(2), c = 11.632(2) A and  $d_{calcd} = 1.468$  g cm<sup>-3</sup> for Z = 4 ( $C_6H_8N_2O_2 \cdot HCl$ , M = 176.60). The systematic absences indicated either space group P2<sub>1</sub>cn or Pmcn. The centric space group Pmcn was assumed and the solution and refinement of the structure proceeded satisfactorily. All atoms, except for 4 methyl hydrogen atoms, lie on the mirror plane. The hydrogen atom on O (2) is hydrogen bonded to the chloride ion (O...Cl distance 2.91 A, O-H...Cl angle 176°).

The intensity data were measured on a Hilger-Watts diffractometer (Ni-filtered CuK $\alpha$  radiation,  $\theta$ -2 $\theta$  scans, pulse height discrimination). The size of the crystal used for data collection was approximately  $0.12 \times 0.12 \times 0.4$  mm; the data were corrected for absorption ( $\mu = 38.8 \text{ cm}^{-1}$ ). Of the 891 accessible reflections for  $\theta < 76$  °C, 766 were considered to be observed. The structure was solved by a multiple solution procedure7 and was refined by full matrix least squares. All hydrogen atoms were located on a difference map calculated after anisotropic refinement of the heavier atoms. In the final refinement, anisotropic thermal parameters were used for the heavier atoms and isotropic temperature factors were used for the hydrogen atoms. The hydrogen atoms were included in the structure factor calculations but their parameters were not refined. The final discrepancy indices are R = 0.042 and wR =0.048 for the 766 observed reflections. There were no peaks greater than  $\pm$  0.2 e A<sup>-3</sup> on the final difference map. This is the first report of norzooanemonin isolated from a hydroid. It has been isolated previously from the Caribbean gorgonian, Pseudopterogorgia americana 6. In both cases it appears to be accompanied by trigonelline and homarine. Each of these betaines is widely distributed in marine invertebrates 4, 5.

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## (+)-Abscisic acid, a metabolite of the fungus Cercospora rosicola

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Summary. The first isolation of (+)-abscisic acid from a fungus, Cercospora rosicola, is reported.

(+)-Abscisic acid (ABA, I) is an important growth regulator in higher plants, and has been isolated from a wide variety of angiosperms and gymnosperms, from a fern, a horsetail, and a moss<sup>2</sup>. It has never been identified in liverworts, algae or fungi<sup>2b</sup>. A single report of abscisic acid in *Penicillium italicum* showed that most probably

the fungus had taken the compound up from its host<sup>3</sup>. We wish to report here what we believe to be the first unambiguous evidence of (+)-ABA formed as a secondary metabolite by a fungus. During the screening of the genus *Cercospora* (Deuteromycetes) for secondary metabolites<sup>4</sup>, we found that a strain<sup>5</sup> of *Cercospora rosicola* Passerini,